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FILE 'CA' ENTERED AT 18:01:32 ON 18 JUL 2005

L1 1508 S (REACTANT OR FEED OR VAPOR) (4A) (PULSE OR PLUG OR PULSES SHAPE OR
SEGMENT)
L2 142 S L1(8A) (DETECT? OR DETERMIN? OR ANALY? OR MEASUR? OR MONITOR? OR
ESTIMAT? OR EVALUAT? OR SENSE# OR SENSING OR SENSOR OR PROBE# OR
PROBING OR QUANTITAT? OR QUANTIF? OR ASSESS? OR EXAMIN?)
L3 27798 S (REACTANT OR FEED OR VAPOR) (4A) (USE OR DEPLET? OR EMPTY OR
DECREAS?)
L4 1209 S L3(8A) (DETECT? OR DETERMIN? OR ANALY? OR MEASUR? OR MONITOR? OR
ESTIMAT? OR EVALUAT? OR SENSE# OR SENSING OR SENSOR OR PROBE# OR
PROBING OR QUANTITAT? OR QUANTIF? OR ASSESS? OR EXAMIN?)
L5 134 S L2,L4 AND (CVD OR VAPOR DEPOSIT? OR LAYER DEPOSIT? OR ALD)
L6 124 S L5 NOT PY>2002
L7 4 S L5 NOT L6 AND PATENT/DT
L8 128 S L6-7

=> d bib,ab 1-128 18

L8 ANSWER 31 OF 128 CA COPYRIGHT 2005 ACS on STN
AN 130:160934 CA
TI Computer controlled **vapor deposition** processes
IN Lemelson, Jerome
PA USA
SO U.S., 15 pp.
PI US 5871805 A 19990216 US 1996-628088 19960408
PRAI US 1996-628088 19960408
AB A method for computerized control of **vapor deposition** processes,
including **CVD** and electron beam phys. **vapor deposition** processes, **uses**
optical imaging **sensors** and/or laser interferometers or IR ellipsometers
focused on the substrate being coated or on a nearby test blank to
provide information which is computer analyzed to yield optimum control
points for the coating process. A method is also disclosed for shaping
or contouring one or more surfaces of an object(s) using the techniques
given here.

L8 ANSWER 52 OF 128 CA COPYRIGHT 2005 ACS on STN
AN 126:109343 CA
TI The adsorption of silane, disilane and trisilane on polycrystalline
silicon: a transient kinetic study
AU Weerts, W. L. M.; de Croon, M. H. J. M.; Marin, G. B.
CS Lab. Chem. Technol., Eindhoven Univ. Technol., Eindhoven, 5600 MB, Neth.
SO Surface Science (1996), 367(3), 321-339
AB The adsorption of silane, disilane, and trisilane on polycryst. Si was
studied by using temporal **anal.** of products (TAP) following on admission
of a **reactant pulse** at 300-1000 K and at pressures typical for low-
pressure chem. **vapor deposition**. At ≤ 650 K, a slow adsorption process
is operative for the 3 silanes. A quant. description of the adsorption
in this temp. range is possible with a mechanism based on an insertion
reaction of the silanes into surface H bonds. At > 650 K, a much faster
mode of adsorption is obsd., which for the higher silanes is accompanied

by silane formation.. Homogeneous gas-phase reactions can be excluded. Silane adsorption at > 810 K can be described quant. with a dual-site adsorption mechanism.

L8 ANSWER 54 OF 128 CA COPYRIGHT 2005 ACS on STN

AN 125:181818 CA

TI Modeling TiN deposition for control of **CVD**

AU Gevelber, Michael; Can Deniz, M.; Liu, Rujiang; Sumitra, Edward
CS Manufg. Eng., Boston Univ., Boston, MA, 02215, USA

SO Proceedings - Electrochemical Society (1996), 96-5 (Chemical Vapor Deposition), 157-162

AB A lumped, nonlinear model was developed for TiN deposition. **Anal.** indicates that **reactant depletion** significantly **dets.** the characteristics of the mass transport regime and has important implications for developing a closed loop control system.

L8 ANSWER 60 OF 128 CA COPYRIGHT 2005 ACS on STN

AN 124:245584 CA

TI Dynamic rate and thickness metrology during poly-Si rapid thermal chemical **vapor deposition** from SiH₄ using real time in situ mass spectrometry

AU Tedder, L. L.; Rubloff, G. W.; Cohaghan, B. F.; Parsons, G. N.
CS NFS Engineering Res. Center for Advanced Electronic Materials Processing, North Carolina State Univ., Raleigh, NC, 27695-7920, USA

SO Journal of Vacuum Science & Technology, A: Vacuum, Surfaces, and Films (1996), 14(2), 267-70

AB Real-time in situ mass spectrometry was applied to poly-Si rapid thermal **CVD** (RTCVD) (from SiH₄) on thermally grown SiO₂ as a way to det. film thickness at the end of the process and to infer dynamic deposition rate during the process for run-to-run and real-time control applications. Monitoring process ambient at 5 torr is achieved using 2-stage differential pumping of a sampling aperture in the exhaust stream, and a rapid response time (~1 s for a ~ 30 s process cycle) allows for real time **sensing** of reactant input, product generation, and **reactant depletion**. Active mass spectrometric sampling of the reaction byproduct (H₂ generated by SiH₄ decompn.) provides a monitor of the total reaction/deposition rate during poly-Si RTCVD in the range 550-850°. Product generation as a function of temp. is readily distinguished from reactant cracking fragments by spectral anal. A well-defined monotonic correlation between the time-integrated H₂⁺ product signal and the poly-Si film thickness, detd. ex situ by single-point interferometry (Nanometrics), demonstrates that the integrated mass spectrometric signal can provide real-time thickness metrol. The time-dependence of product and reactant signals provides a real-time indication of detailed equipment behavior during the process.

L8 ANSWER 68 OF 128 CA COPYRIGHT 2005 ACS on STN

AN 122:120112 CA

TI Use of residual gas analysis in low pressure semiconductor process reactors

AU Reath, Mark; Brannen, James; Bakeman, Paul; Lebel, Richard

CS IBM Technology Products, Essex Junction, VT, 05452, USA

SO Proceedings - Institute of Environmental Sciences (1993), 39TH(VOL. 1),

119-23

AB Residual gas anal. (RGA) was used for trouble-shooting of TEOS and tungsten **CVD** processes. In each process, RGA identified reactor impurity sources later proven to be the root cause of film defects and foreign material deposition. RGA verified the effectiveness of modified reactor hardware and operating procedures.

L8 ANSWER 113 OF 128 CA COPYRIGHT 2005 ACS on STN

AN 102:88083 CA

TI Vapor growth with monitoring

IN Nisizawa, Junichi; Fukase, Masaaki

PA Semiconductor Research Foundation, Japan

SO U.S., 20 pp. Cont. of U.S. Ser. No. 161,980, abandoned.

PI US 4479845 A 19841030 US 1982-388198 19820614

PRAI JP 1979-151206 A 19791120

AB The app. for vapor growth of Si, and the in situ monitoring device detg. the progress of the growth, are described. The device can also be used to grow Si by epitaxy with desired doping profile. Various portions of the substrate can be used as sampling points as well as diffusion source for supplying dopants.

L8 ANSWER 125 OF 128 CA COPYRIGHT 2005 ACS on STN

AN 71:14506 CA

TI **Vapor deposition** techniques

AU Holzl, Robert A.

CS San Fernando Lab., Pacoima, CA, USA

SO Techniques of Metals Research (1968), 1(Pt. 3), 1377-405

AB A survey of chem. **vapor deposition** practice, including the equipment and methods used and the substances deposited, is given, together with notes on practical applications and the chem. reactions involved. Items covered include the **detn.** of reaction feasibility, the **use** of halide **reactants**, bulb and feed systems, design data, and current practice for the **vapor deposition** by pyrolysis or redn. techniques of metals and other materials. 44 references.

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